

## XXVII. Solution of Schrödinger Wave Equation for Propagation of an Electron

Given an electron of mass,  $m_e$ , velocity,  $v_e$ , kinetic energy of 1 eV

By Quantum Mechanics. it has an associated de Broglie wavelength,  $\lambda_e$ , and wavenumber  $k_0$

Planck's Constant:  $h := 6.626 \cdot 10^{-34} \text{ J}\cdot\text{s}$

**Given:**  $m_e := 9.10938 \cdot 10^{-31} \text{ kg}$      $v := 1 \cdot 10^3 \frac{\text{m}}{\text{s}}$      $T := 1 \text{ eV}$      $\lambda_e := \frac{h}{\sqrt{2 \cdot m_e \cdot T}} = 12.265 \cdot A$

Consider a monochromatic E Field plane wave associated with an electron which propagates in an isotropic and homogeneous medium:  $E(r, t) = E_0 \cdot e^{[i \cdot (k \cdot r - \omega t)]}$

Associated with this electron is a wavenumber,  $k_e$ , amplitude,  $A$

$$k_e := 8.637 \cdot 10^6 \cdot \frac{1}{\text{m}} \quad A_{\text{m}} := \frac{1}{\sqrt{a \cdot \sqrt{\pi}}} \quad h_{\text{bar}} := \frac{h}{2 \cdot \pi}$$

$$A_m := A \cdot \sqrt{m}$$

$$eV := 1.602 \cdot 10^{-19} \text{ J}$$

The electron has Kinetic Energy:  $E := \frac{h_{\text{bar}}^2 \cdot k_e^2}{2 \cdot m_e}$      $E = 4.554 \times 10^{-25} \text{ J}$     **Electron Frequency:**  $\omega = 2\pi \cdot f$

**Equation of Traveling Wave:**

$$\psi(x) = A \cdot e^{i \cdot (k \cdot x - \omega t)} + B \cdot e^{-i \cdot (k \cdot x + \omega t)}$$

To solve the one-dimensional Schrödinger equation for a free particle of mass  $m$  moving with velocity  $v$ , we can proceed as follows:

### Solve Schrödinger's Wave Equation for the Quantum Wavefunction, $\Psi(x, t)$

$$\frac{-\hbar^2}{2 \cdot m_e} \cdot \frac{\partial^2}{\partial x^2} \Psi = i \cdot \hbar \cdot \frac{\partial}{\partial t} \Psi$$

$$\Psi(x, t) = \int_{-\infty}^{\infty} \psi(k, x, t) dk$$

$$\Psi(x, t) = \frac{A \cdot a}{\sqrt{2\pi}} \int_{-\infty}^{\infty} \exp \left[ \frac{-1}{2} a^2 \cdot (k - k_0)^2 + i \cdot k \cdot x - \frac{i \cdot h_{\text{bar}} \cdot t}{2 \cdot m_e} \cdot k^2 \right] dk$$

Evaluate the Wavefunction over the Space and Time Region:

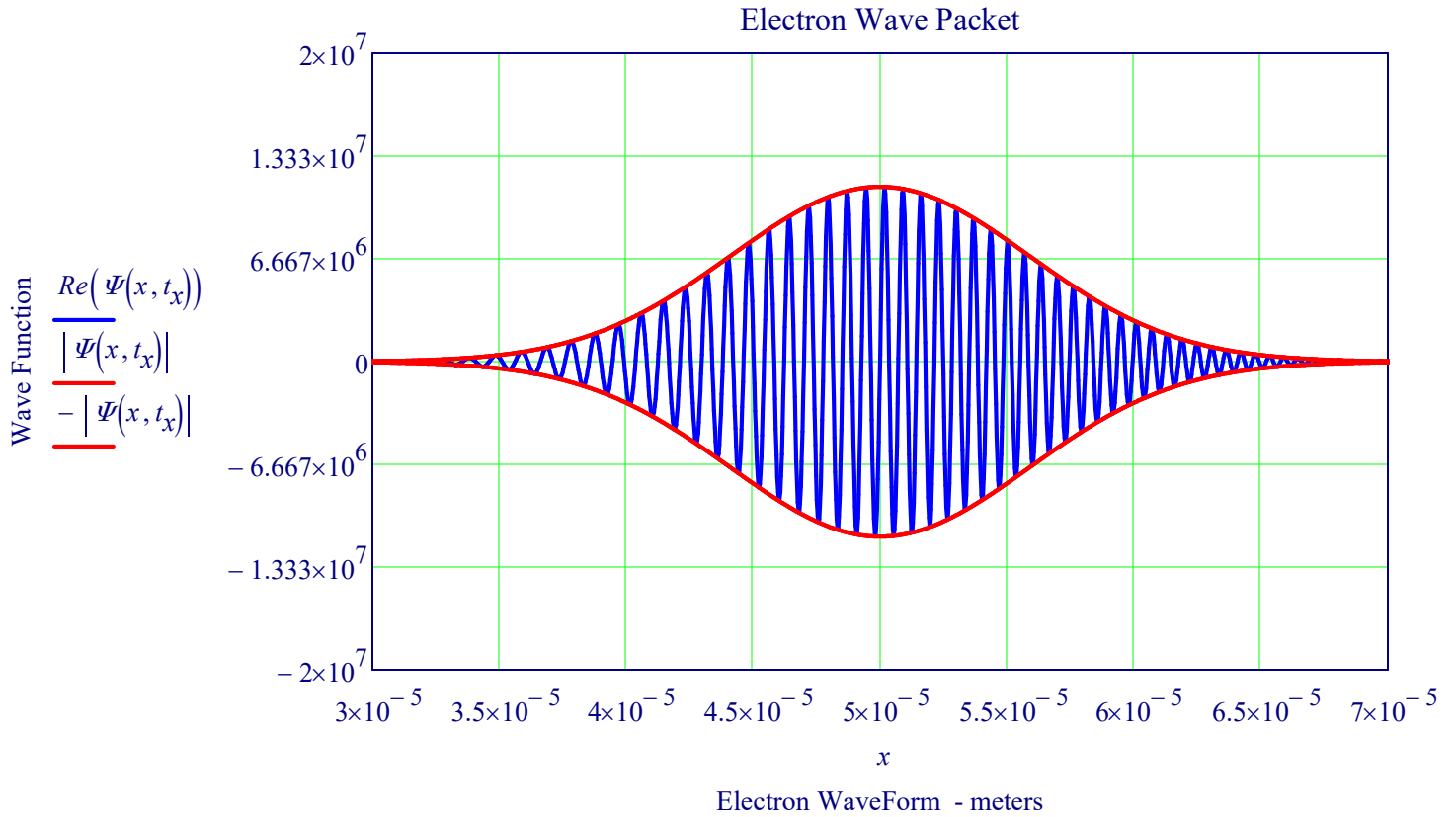
$$a := 1 \cdot \mu\text{m} \quad t_x := 50 \text{ ns}$$

**Solution for  $\Psi(x, t)$ :**

$$\Psi_{\text{m}}(x, t) := \frac{A_m}{\sqrt{1 + \frac{i \cdot h_{\text{bar}} \cdot t}{m_e \cdot a^2}}} \cdot \exp \left[ \frac{- \left( x^2 - 2 \cdot i \cdot a^2 \cdot k_e \cdot x + \frac{i \cdot h_{\text{bar}} \cdot t}{2 \cdot m_e} \cdot k_e^2 \cdot a^2 \right)}{2 \cdot a^2 \cdot \left( 1 + \frac{i \cdot h_{\text{bar}} \cdot t}{m_e \cdot a^2} \right)} \right]$$

## Plot Wavefunction $\Psi(x,t)$ over Distance Range, x

**Distance Range:**  $x := 10^{-5} \cdot 2m, 10^{-5} \cdot 2m + \left( \frac{10^{-5} \cdot 8 \cdot m - 10^{-5} \cdot 2m}{2000} \right) .. 10^{-5} \cdot 8m$



# Numerical Schrödinger Equation Solutions for 3-D Harmonic Oscillator

## Parameters:

Reduced mass:  $\mu := 1$     Angular momentum:  $L := 0$     Integration limit:  $r_{max} := 6$      $E := 7.5$      $L := 0$   
 Force constant:  $k := 1$      $r := 0, 0.01 .. r_{max}$

## Solve Schrödinger's equation numerically. Use Mathcad's ODE solve block:

Given

$$\frac{-1}{2 \cdot \mu} \cdot \frac{d^2}{dr^2} \Psi(r) - \frac{1}{r \cdot \mu} \cdot \frac{d}{dr} \Psi(r) + \left[ \frac{L \cdot (L + 1)}{2 \cdot \mu \cdot r^2} + \frac{1}{2} \cdot k \cdot r^2 \right] \cdot \Psi(r) = E \cdot \Psi(r)$$

$$\Psi(.001) = .1 \quad \Psi'(.001) = .1$$

$$\Psi := \text{Odesolve}(r, r_{max})$$

Energy guess:  $E \equiv 7.5$

$$\text{Normalize the wavefunction: } \Psi(r) := \left( \int_0^{r_{max}} \Psi(r)^2 \cdot 4 \cdot \pi \cdot r^2 \, dr \right)^{\frac{-1}{2}} \cdot \Psi(r)$$

